

SYNTHESIS AND STEREOCHEMICAL ASSIGNMENT OF THE C₁-C₁₀ FRAGMENT OF NYSTATIN A₁

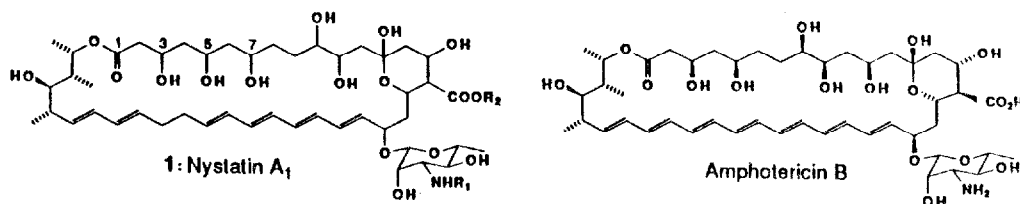
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Summary. Compound **8** representing the C₁-C₁₀ fragment of nystatin A₁ has been synthesized from nystatin A₁ by degradation and by total synthesis. Comparison of the two samples revealed the 3*R*, 5*R*, 7*R* stereochemistry for the three nystatin A₁ centers.

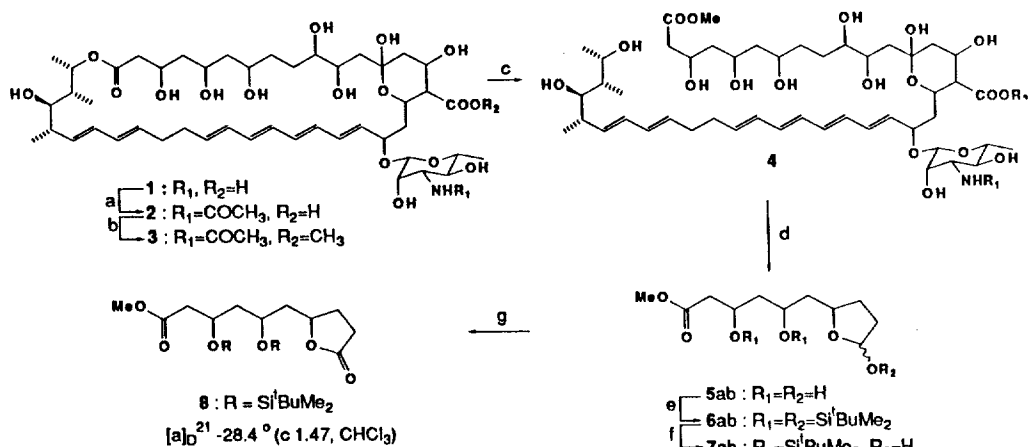
Nystatin A₁ (**1**)¹ is a clinically used polyene macrolide antibiotic of considerable importance. Despite its widespread use in medicine, however, its stereochemistry remains only partially known¹. Our efforts in the area of polyene macrolide antibiotics, which recently culminated in the total synthesis² of amphotericin B (**2**) and its aglycon, are now directed towards the total synthesis and structural elucidation of nystatin A₁ (**1**). A recent communication³ regarding the relative stereochemistry of the hydroxyl groups of the C₁-C₁₀ fragment of nystatin A₁ (**4**) prompted us to report our investigations in this area which resulted in the synthesis and absolute stereochemical assignment of this fragment.

The C₁-C₁₀ fragment of nystatin A₁ (**1**) was excised from the natural product by the sequence depicted in **Scheme 1**. Thus, nystatin A₁ (**2**)⁴ was monoacetylated in methanol with acetic anhydride to give the acetamide **2** which was then methylated with diazomethane leading to methyl ester **3** (78% from **1**). Basic methanolysis of **3** under carefully controlled conditions led to compound **4** (30% yield

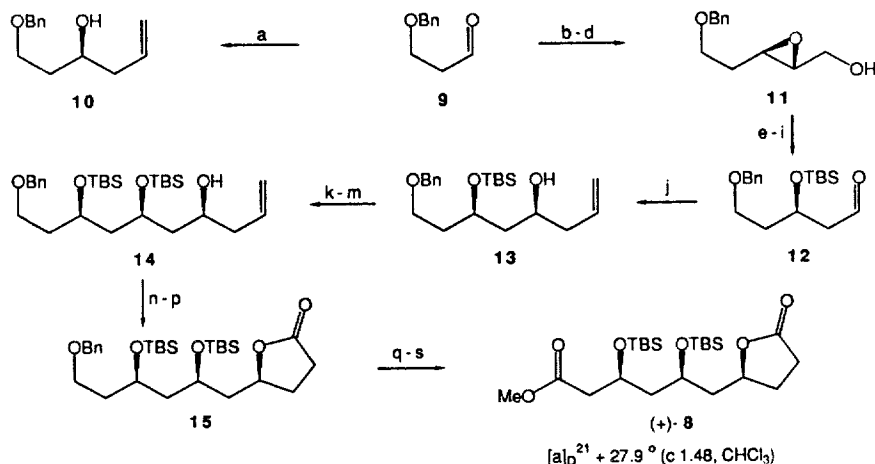


plus 26% recovered starting material) which was then cleaved at the 1,2-diol site with NaIO_4 in aqueous methanol furnishing the trihydroxy compound **5ab** (43%) as a mixture of lactol epimers (ca 2.5:1 by ^1H NMR). In addition to **5ab**, a second major product was obtained in this reaction which was presumed to be the other expected fragment of the molecule, although it has not yet been fully characterized. Compound **5ab** was then persilylated with $^t\text{BuMe}_2\text{SiCl}$ -imidazole to afford trisilyl ether **6ab** which was selectively converted to lactol **7ab** (mixture of epimers; ca 2.5:1 by ^1H NMR) by exposure to mildly acidic conditions (AcOH -THF- H_2O , 66%). Finally, oxidation of **7ab** with pyridinium chlorochromate afforded γ -lactone **8**⁵ in 96% yield [$[\alpha]_{\text{D}}^{21}$ -28.4° (c 1.47, CHCl_3)]. In order to determine the relative and absolute stereochemistry of **8** synthetic studies were undertaken. In considering likely stereochemical candidates for **8**, our first suspect was the all-syn-3R,5R,7R compound due to the similarities of nystatin A₁ (**1**) to amphotericin B, although by no means this analogy precluded other possible structures⁶. It was also decided that Brown's allyl (lpc)₂ borane reagents⁷ would be used to provide the necessary asymmetric induction and flexibility in constructing the requisite stereoisomers. However, due to the higher optical purity of the commercially available⁸ to us (-)-reagent at the time, we proceeded initially to build the 3S,5S,7S compound as depicted in **Scheme 2**.

Aldehyde **9** (**Scheme 2**) was initially reacted with the allyl Brown reagent derived from (-)-B-lpc₂Ome and $\text{CH}_2=\text{CHCH}_2\text{MgBr}$ giving hydroxy compound **10** in 80% yield but only in 74% ee. On the other hand standard elaboration of **9** to epoxy alcohol **11** via the Sharpless epoxidation procedure¹⁰ resulted in 77% overall yield and 91% ee⁹. The later sequence with the higher ee was therefore utilized in the synthesis of **8**. Thus, **11** was regioselectively opened to the corresponding 1,3-diol with REDAL¹¹ (95%) and then elaborated to aldehyde **12** by standard chemistry as outlined in **Scheme 2** (82% overall yield). Addition of the allyl Brown reagent derived from (-)-B-lpc₂Ome and $\text{CH}_2=\text{CHCH}_2\text{MgBr}$ then led to compound **13** in 92% yield and 83% diastereomeric excess¹². Careful flash column chromatography produced pure **13** from which pure **14** was generated by (a) protection; (b) ozonolysis- PPh_3 (c) reiteration of the Brown addition process and (d) chromatography (88% overall yield). The conversion of **14** to the desired target **8** for comparison with the degradatively derived material proceeded through intermediate **15** as follows: (a) temporary protection of the alcohol as ethoxyethyl ether (b) hydroboration of the olefin with 9-BBN followed by basic hydrogen peroxide work-up; (c) PCC oxidation of the alcohol with concomitant deprotection-cyclization-oxidation to afford **15** (44% overall yield); (d) hydrogenolysis of the benzyl ether; and (e) oxidation of the resulting primary alcohol, first to the aldehyde (PDC, CH_2Cl_2) and then directly¹³ to the methyl ester **8** (58% overall yield). Synthetic **8** exhibited identical chromatographic and spectroscopic data as degradatively derived **8**, but opposite optical rotation [$[\alpha]_{\text{D}}^{21}$ $+27.9^\circ$ (c 1.48, CHCl_3)]. The CD spectra¹⁴ of the two samples also exhibited opposite cotton effect reinforcing the conclusion of their enantiomeric nature. Therefore, based on the expected stereochemical course of the Sharpless¹⁰ and Brown^{7,6} reactions employed, the 3R,5R,7R absolute stereochemistry is assigned for fragment (-)-**8** and the corresponding nystatin A₁ (**1**) stereogenic centers. Work towards the complete stereochemical assignments and total synthesis of nystatin A₁ is continuing^{14,15}.



Scheme 1. Degradation of nystatin A₁ to compound 8. Reagents and conditions: (a) 2.0 equiv Ac₂O, MeOH, 0 °C, 1 h; (b) excess CH₂N₂, THF-MeOH, 25 °C, 1 h, 78% from 1; (c) 1.2 equiv K₂CO₃, MeOH, 25 °C, 12 h, 30%, plus 26% recovered starting material; (d) 2.0 equiv NaIO₄, MeOH-H₂O(20:1), 25 °C, 30 min, 43%; (e) 5.0 equiv ^tBuMe₂SiCl, imidazole, DMAP, cat., DMF, 25 °C, 12 h, 100%, (f) AcOH-THF-H₂O(3:3:1), 0 - 25 °C, 1 h, 66% (g) 2.5 equiv PCC, 4A molecular sieve, CH₂Cl₂, 25 °C, 1 h, 96%.



Scheme 2. Synthesis of 3S,5S,7S C₁-C₁₀ fragment (+)-8 from 3-benzyloxypropanal 9. Reagents and conditions: (a) 1.1 equiv *lpc*₂BCH₂CH=CH₂ (from (-)-methoxydiisopinocampheylborane and BrMgCH₂CH=CH₂), Et₂O, -78 - 25 °C, 2 h; 30% H₂O₂-3N-NaOH, 50 °C, 2 h, 74%; (b) 1.0 equiv (EtO)₂P(O)CH₂COOEt, ^tBuOK, THF, -78 °C, 30 min, 91%; (c) 2.5 equiv DIBAL, PhCH₃, -78 - -45 °C, 2 h, 95%; (d) 0.12 equiv L-(-)-DET, 0.10 equiv Ti(O-*i*Pr)₄, 2.0 equiv ^tBuOOH, 20% wt 4A molecular sieve, 0.1 M in CH₂Cl₂, -20 °C, 12 h, 89%; (e) 1.5 equiv REDAL, THF, -20 °C, 12 h, 95%; (f) 1.1 equiv ^tBuCOCl, pyridine, -20 °C, 10 min, 89%; (g) 1.2 equiv ^tBuMe₂SiCl, imidazole, DMAP, cat., DMF, 25 °C, 12 h, 98%; (h) 2.5 equiv DIBAL, THF-hexane, -78 °C, 1 h, 96%; (i) 1.1 equiv (COCl)₂, 2.5 equiv DMSO, 5.0 equiv Et₃N, CH₂Cl₂, -78 - 0 °C, 1 h, 98%; (j) same as a, 92%; (k) same as g, 98%; (l) O₃, CH₂Cl₂-MeOH(100:1), -78 °C, ; 2.0 equiv Ph₃P, 25 °C, 2 h, 94%; (m) same as a, 95%; (n) 1.2 equiv CH₂=CHOCH₂CH₃, PPTS, cat., CH₂Cl₂, 25 °C, 30 min, 71%; (o) 1.5 equiv 9-BBN, THF, 0 - 25 °C, 3 h, 92%; (p) 5.0 equiv PCC, 4A molecular sieve, 25 °C, 1 h, 68%; (q) H₂, Pd(OH)₂, cat., THF, 25 °C, 3 h, 94%; (r) 4.0 equiv PDC, 4A molecular sieve, CH₂Cl₂, 25 °C, 1.5 h, 78%; (s) 6.0 equiv PDC, 6.0 equiv MeOH, DMF, 25 °C, 24 h, 78%.

References and Notes

1. (a) Hazen, E.L.; Brown, R. *Science* **1950**, *112*, 423. (b) Chong, C.N.; Rickards, R.W. *Tetrahedron Lett.* **1970**, 5145. (c) Borowski, E.; Zielinski, J.; Falkowski, L.; Ziminski, T.; Golik, J.; Kolodziejczyk, P.; Jereczek, E.; Gdulewicz, M.; Shenin, Y.; Kotienko, T. *Tetrahedron Lett.* **1971**, 685. (d) Dutcher, J.D.; Walters, D.R., Wintersteiner, O. *J. Org. Chem.* **1963**, *28*, 995. (e) Von Saltza, M.; Dutcher, J.D.; Reid, J.; Wintersteiner, O. *J. Org. Chem.* **1963**, *28*, 999. (f) Jereczek, E.; Sowinski, P.; Zielinski, J.; Borowski, E. *Inst. Nucl. Phys. Cracow. Rep.* **1973**, 232. (g) Iketa, M.; Suzuki, M.; Djerassi, C. *Tetrahedron Lett.* **1987**, 3745.
2. Nicolaou, K.C.; Chakraborty, T.K.; Ogawa, Y.; Daines, R.A.; Simpkins, N.S.; Furst, G.T. *J. Am. Chem. Soc.* **1988**, *110*, 4660. Nicolaou, K.C.; Daines, R.A.; Uenishi, J.; Li, W.S.; Papahatjis, D.P.; Chakraborty, T.K. *J. Am. Chem. Soc.* **1988**, *110*, 4672. Nicolaou, K.C.; Daines, R.A.; Chakraborty, T.K.; Ogawa, Y. *J. Am. Chem. Soc.* **1988**, *110*, 4685. Nicolaou, K.C.; Daines, R.A.; Ogawa, Y.; Chakraborty, T.K. *J. Am. Chem. Soc.* **1988**, *110*, 4696.
3. Lancelin, J.-M.; Paquet, F.; Beau, J.-M. *Tetrahedron Lett.*, **1988**, *29*, 2827.
4. We thank Dr. C. Cimarusti, E.R. Squibb, P.O. Box 4000, Princeton, NJ 08054, for generous samples of nystatin A₁ (**1**).
5. R_f = 0.36 (silica, 20% EtOAc in petroleum ether); ¹H NMR (500 MHz, CDCl₃) δ 4.61 (m, 1 H, H-7), 4.21 (m, 1 H, H-3), 3.93 (m, 1 H, H-5), 3.64 (s, 3 H, COOCH₃), 2.51-2.47 (m, 3 H, H-2, H-9), 2.42 (dd, J = 11.5, 5.5 Hz, 1 H, H-2), 2.31 (m, 1 H, CH₂), 1.96-1.72 (m, 5 H, CH₂), 0.86, 0.83 (singlets, 9 H each, *t*Bu), 0.06, 0.04, 0.03, 0.02 (singlets, 3 H each, SiMe₂). IR (neat) ν_{max} 1783 (γ-lactone), 1744 cm⁻¹ (COOMe); HRMS Calcd for C₂₃H₄₆O₆Si₂: 474.2830, found: 474.2841.
6. This caution was recently emphasized and justified by S.L. Schreiber and Goulet in their elegant studies on mycoticin A and B: (a) Schreiber, S.L.; Goulet, M.T. *Tetrahedron Lett.* **1987**, *28*, 6001.
7. Brown, H.C.; Jadhav, P.K. *J. Am. Chem. Soc.* **1983**, *105*, 2092; Jadhav, P.K.; Bhat, K.S.; Perumal, T.; Brown, H.C. *J. Org. Chem.* **1986**, *51*, 432.
8. Aldrich Co; [α]_D²⁵ -73.7° (c=1, THF) for (-)-B-methoxydiisopinocampheylborane.
9. Enantiomeric excess (% ee) was determined by ¹⁹F NMR spectroscopy using the Mosher ester: Dale, J.A.; Dull, D.L.; Mosher, H.S. *J. Org. Chem.*, **1969**, *34*, 2543.
10. Gao, Y.; Hanson, R.M.; Klunder, J.M.; Ko, S.Y.; Masumune, H.; Sharpless, K.B. *J. Am. Chem. Soc.* **1987**, *109*, 5765.
11. Finan, J.M.; Kishi, Y. *Tetrahedron Lett.*, **1982**, *23*, 2719.
12. The diastereomeric excess was determined by ¹H NMR spectroscopy. Based on this value (83%), a 91% ee was calculated for the asymmetric induction of the second stereogenic center in **13**.
13. O'Connor, B.; Just, G. *Tetrahedron Lett.* **1987**, *28*, 3235.
14. Degradatively derived **8** showed a positive Cotton effect, whereas the synthetic material exhibited a negative Cotton effect. We thank Professor K. Nakanishi and W.T. Wiesler of Columbia University for the measurements of the CD spectra.
15. All new compounds exhibited satisfactory spectroscopic and/or analytical and exact mass data.
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